

SYNTHESIS OF 9-(3-DEOXY- AND 2,3-DIDEOXY-3-FLUORO- β -D-XYLOFURANOSYL)GUANINES AS POTENTIAL ANTIVIRAL AGENTS

Frédéric PUECH, Gilles GOSSELIN and Jean-Louis IMBACH*

Université des Sciences et Techniques du Languedoc, Laboratoire de Chimie Bio-organique associé au CNRS, Place E. Bataillon, 34060 Montpellier, France.

Abstract : The first synthesis of the title compounds **8** and **10** was accomplished by a multi-step approach involving prior preparation of a suitably protected fluorosugar **6**.

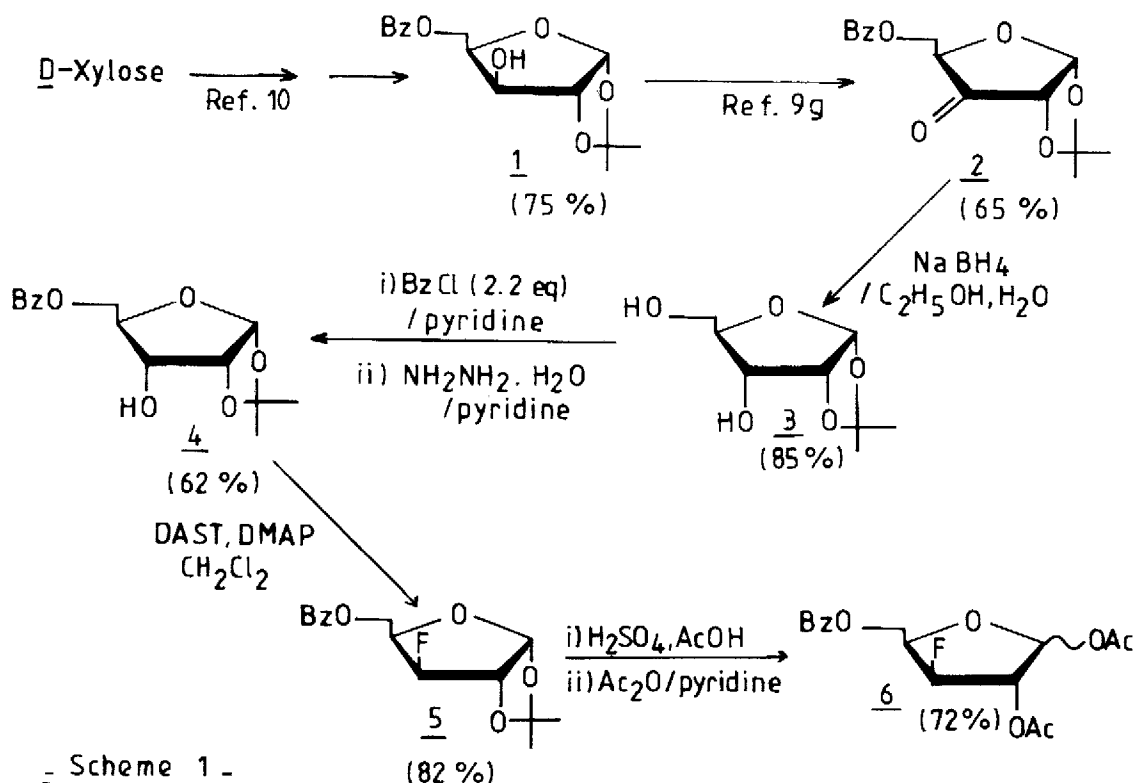
Nucleosides fluorinated at the 2'-position of the sugar moiety have been extensively investigated in the search for anti-viral agents, and some 2'-deoxy- and 2',3'-dideoxy-2'-ara-fluoro analogues actually display potent anti-herpes¹ and anti-human immunodeficiency virus² (anti-HIV) activities, respectively. In contrast, nucleosides fluorinated at their 3'-position have been less explored. For instance, among the analogues with 3'-xylo-fluoro configuration two derivatives (those of adenine³ and cytosine⁴) in the 3'-deoxy series and only one (of adenine⁵) in the 2',3'-dideoxy series have been prepared previously. The recently mentioned inhibitory effects of the latter compound on the replication of HIV⁶ prompted us to synthesize the hitherto unknown 9-(3-deoxy- and 2,3-dideoxy-3-fluoro- β -D-xylofuranosyl)guanine (**8**) and (**10**) in order to evaluate their antiviral activities.

The most appropriate synthetic plan to reach **8** and **10** appeared to first prepare a suitably fluorodeoxy sugar like **6** and to condense it with N²-acetylguanine. After condensation and selective 2'-deacetylation the resulting key nucleoside intermediate **7** could be fully deprotected, either directly or after 2'-deoxygenation via a Barton-type reaction,⁷ to give **8** and **10**, respectively.

To date there have been two indirect approaches to the synthesis of 3-deoxy-3-fluoro- β -D-xylofuranosyl sugars, involving nucleophilic displacement of the sulfonate group from 1,2:5,6-di-O-isopropylidene-3-O-toluene-p-sulfonyl- α -D-allofuranose⁸ and action of potassium hydrogen fluoride on methyl 2,3-anhydro-5-O-benzyl- β -D-ribofuranoside.^{3a} In order to make these fluoro sugars more accessible we have synthesized 1,2-di-O-acetyl-5-O-benzoyl-3-deoxy-3-fluoro-D-xylofuranose (**6**) from D-xylose.

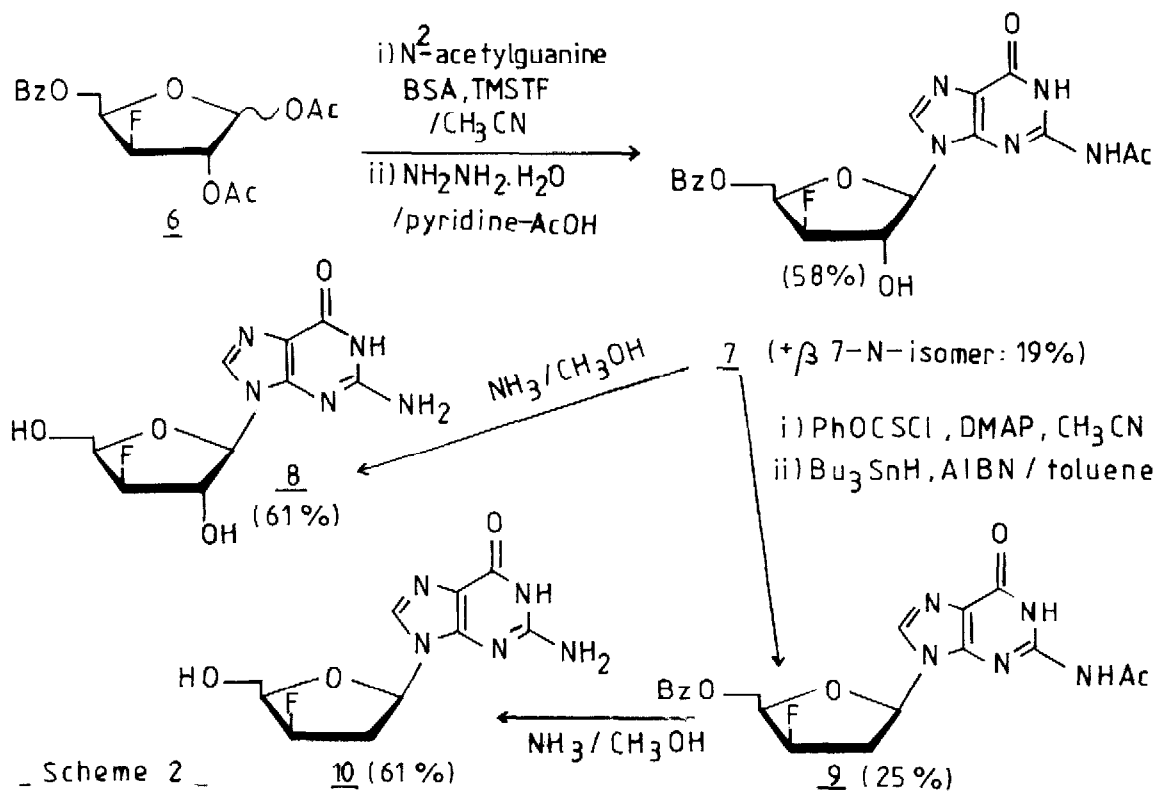
Thus, D-xylose was first conveniently converted on a large scale into 5-O-benzoyl-1,2-O-isopropylidene- α -D-erythro-pentofuranose-3-ulose (**2**)⁹ according to the most effective published procedure.^{10, 9g} The reduction of **2** with sodium borohydride in aqueous ethanol gave exclusively 1,2-O-isopropylidene- α -

-D-ribofuranose (3)^{9e,11} which on reaction with an excess of benzoyl chloride and subsequent treatment with hydrazine hydrate afforded 5-O-benzoyl-1,2-O-isopropylidene- α -D-ribofuranose (4).¹² Substitution with inversion of configuration of the 3'-hydroxy group of 4 by a fluorine atom was performed using diethylaminosulfur trifluoride (DAST). Fluorination of 4 under conditions usual in the hexose series,¹³ afforded 5-O-benzoyl-1,2-O-isopropylidene-3-deoxy-3-fluoro- α -D-xylofuranose (5) in good yield. After selective removal of the 1,2-O-isopropylidene group from 5 and acetylation, the required sugar 6 was obtained (Scheme 1).



Condensation of 6 with silylated N^2 -acetylguanine, separation of the resulting β 9-N,7-N isomers and full deacylation to give ultimately the desired fluoronucleoside 8 were effected under conditions similar to those used for the synthesis of β -D-xylofuranosylguanine in our earlier paper.¹⁴ The isolated intermediate 7 on reductive 2'-deoxygenation⁷ followed by a deacylation afforded the second desired nucleoside 10 (Scheme 2).

Structural assignments for the compounds 1-10 were based on their physicochemical properties.¹⁵ Unfortunately, the title compounds 8 and 10 were shown to be ineffective against various DNA and RNA viruses in primary rabbit kidney, Hela and Vero cell cultures. They were also not active against HIV-1 in MT4 cell cultures.



Further data and studies on the use of the fluorodeoxy sugar **6** in the synthesis of other nucleoside analogues will be reported in a full paper.

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References and Notes.

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15. For previously described compounds 1-4 our data were in accord with literature values. Hitherto unknown compounds 5-10 were characterized by ¹H and ¹⁹F n.m.r., u.v., and mass spectrometry. *Selected spectroscopic data*: For 8: ¹H n.m.r. (Me₂SO-d₆) δ (relative to Me₂SO-d₅ set at 2.51 ppm) 10.7 (1H, br s, NH-1), 7.63 (1H, s, H-8), 6.6 (2H, br s, NH₂), 6.2 (1H, br s, OH-2'), 5.72 (1H, d, H-1'; J = 2.1 Hz), 5.05 (1H, d, H-3'; J_{3',F} = 51.7 Hz), 5.03 (1H, t, OH-5'), 4.60 (1H, dm, H-2'; J_{2',F} = 15.8 Hz), 4.23 (1H, m, H-4'; J_{4',F} = 26.0 Hz), 3.70 (2H, m, H-5',5"); ¹⁹F n.m.r. (Me₂SO-d₆) δ (relative to CFCl₃) -200.16 (ddd, F-3'; J_{F,3'} = 52.0 Hz, J_{F,2'} = 16.0 Hz, J_{F,4'} = 29.0 Hz); m/z (f.a.b. > 0) (glycerol) 152 (BH₂), 286 (M+H), 378 (M+G+H), 571 (2M+H); For 10: ¹H n.m.r. (Me₂SO-d₆) δ 10.64 (1H, s, NH-1), 7.64 (1H, s, H-8), 6.5 (2H, br s, NH₂), 6.14 (1H, dd, H-1'; J_{1',2'} = 7.5 Hz, J_{1',2'} = 2.0 Hz), 5.36 (1H, ddd, H-3'; J_{3',F} = 53.2 Hz, J_{3',2'} = 4 Hz, J_{3',4'} = 2.7 Hz), 4.98 (1H, t, OH-5'; J = 5.6 Hz), 5.05 (1H, ddt, H-4'; J_{4',F} = 29.9 Hz, J_{4',(5',5'')} = 6.3 Hz, J_{4',3'} = 2.5 Hz), 3.8-3.6 (2H, m, H-5',5''), 2.84 (1H, dddd, H-2'', J_{2'',F} = 42.3 Hz, J_{2'',2'} = 15.8 Hz, J_{2'',1'} = 7.5 Hz, J_{2'',3'} = 4.5 Hz), 2.58 (1H, ddd, H-2'; J_{2',F} = 25.1 Hz, J_{2',2''} = 16 Hz, J_{2',1'} = 2 Hz); ¹⁹F n.m.r. (Me₂SO-d₆) δ -191.48 (dddd, F-3'; J_{F,3'} = 54.0 Hz, J_{F,2''} = 42.5 Hz, J_{F,4'} = 30.0 Hz, J_{F,2'} = 23.8 Hz); m/z (f.a.b.>0) (glycerol) 152 (BH₂), 270 (M+H).